

Total Synthesis of Verruculogen and Fumitremorgin A Enabled by Ligand-Controlled C–H Borylation

Yu Feng,[‡] Dane Holte,[‡] Jochen Zoller, Shigenobu Umemiya, Leah R. Simke, and Phil S. Baran*

Department of Chemistry, The Scripps Research Institute, 10550 North Torrey Pines Road, La Jolla, California 92037, United States

Supporting Information

ABSTRACT: Verruculogen and fumitremorgin A are bioactive alkaloids that contain a unique eight-membered endoperoxide. Although related natural products such as fumitremorgins B and C have been previously synthesized, we report the first synthesis of the more complex, endoperoxide-containing members of this family. A concise route to verruculogen and fumitremorgin A relied not only on a hydroperoxide/indole hemiaminal cyclization, but also on the ability to access the seemingly simple starting material, 6-methoxytryptophan. An iridium-catalyzed C-H borylation/Chan-Lam procedure guided by an N-TIPS group enabled the conversion of a tryptophan derivative into a 6-methoxytryptophan derivative, proving to be a general way to functionalize the C6 position of an N.C3-disubstituted indole for the synthesis of indolecontaining natural products and pharmaceuticals.

umitremorgin A $(1, \text{ Figure } 1)^1$ and vertuculogen $(2)^2$ stand alone as the only family of alkaloids that harbor an eight-membered endoperoxide.³ Indeed, of the five-, six-, and eight-membered rings embedded in these hexacyclic alkaloids, the peroxide bridge is perhaps their most astounding feature. The juxtaposition of this classic oxidant with two nearby prenyl groups and a readily oxidizable 6-methoxyindole residue add to the allure of this 40-year-old unanswered synthetic puzzle. Aside from their exotic structure, these polycyclic tryptophanbased natural products were first identified due to their tremorinducing activity in mice,^{1,2} and the fumitremorgins display potent activity against multi-drug resistant (MDR) cancer cell lines⁴ and also against HIV by a similar mechanism.⁵ Although family members such as fumitremorgins B and C as well as verruculogen-TR2 have been previously synthesized,⁶ the more complex, endoperoxide-containing compounds 1 and 2 have not yet been made. In this Communication, we present concise total syntheses of 1 and 2 that hinge upon a scalable solution to the synthesis of 6-substituted tryptophan and a diastereoselective peroxide-forming ring closure.

A suitable retrosynthetic precursor to fumitremorgin A (1) and verruculogen (2) is 6-methoxytryptophan methyl ester (3), which would ideally be derived from a chiral and inexpensive starting material, L-tryptophan methyl ester (Figure 1). To our surprise, a short, scalable, regio-controlled route to 3 had not been reported. Given the sheer number of tryptamine-based alkaloids with C6 substitution, this is not such an esoteric problem.⁷ The only direct route to 3 from a tryptophan derivative involves a lead tetraacetate oxidation to afford a



Figure 1. Retrosynthetic analysis of fumitremorgin A (1) and vertuculogen (2).

complex mixture of 5-, 6-, and 7-oxytryptophans.^{6,8} Known routes to 3 or its derivatives involve either ring synthesis⁹ or C3-functionalization of 6-methoxyindole,^{10,11} which in turn is formed by ring synthesis^{11a,c,12} (see summary of literature routes in the Supporting Information). To fill this gap in methodology, C–H functionalization logic¹³ was applied to the C6 problem.

The implicit challenge of a strategy seeking to modify only the C6-H bond is evident, given the known propensity of indole to react preferentially at C2,¹⁴ C3,¹⁴ and C7.¹⁵ In fact, it could be argued that the C6 position is the most difficult position to directly functionalize on an indole, even when considering electrophilic aromatic substitution and directed ortho-metalation strategies.¹⁶ C–H borylation was selected as the method of choice to solve this problem, given its demonstrated utility on tryptophan systems^{14b,15d} and recent reports showing ligand control of regioselectivity.¹⁷ Additionally, we postulated that a large blocking group on the indole nitrogen, such as a TIPS group, would be necessary to obtain the desired regioselectivity. Thus, Boc-L-Trp(TIPS)OMe (4, Figure 2) was chosen as our starting point in order to shield the C2 and C7 positions from reacting.^{14a} Initially, reaction of 4 with B_2Pin_2 under catalytic $[Ir(cod)Cl]_2$ and dtbpy (L1) in octane^{14a} led to low reactivity (5% isolated yield) and low regioselectivity (2:1) of the desired C6-borylated tryptophan product 6 (Figure 2A, entry 1). Employing conditions developed for pyrrole-type heteroarenes,^{14b} which involve HBPin under catalytic $[Ir(cod)OMe]_2$ and dtbpy (L1) in MTBE (specific solvent for tryptophan substrates), none of the desired C6-borylated product was obtained (entry 2). As such,

Received: July 9, 2015 Published: August 9, 2015



Figure 2. C–H borylation of indole derivatives for the synthesis of C6-substituted tryptophans. (A) Optimization of the C–H borylation reaction on tryptophan derivative **4** in order to preferentially achieve C6-borylation. (B) Substrate scope involving C3-substituted indoles as well as carbazoles.

optimization of reaction conditions was needed. In order to understand the effect of B₂Pin₂ versus HBPin, entry 1 was repeated using HBPin instead of B₂Pin₂ (entry 3), and entry 2 was repeated using B₂Pin₂ instead of HBPin (entry 4), but neither of these conditions provided product. A solvent change from MTBE to a nonpolar solvent, hexanes,^{14b} gratifyingly resulted in 23% isolated yield of a 2:1 mixture of 6 and 5, with 52% recovered 4 (entry 5). When both HBPin and B₂Pin₂ were added,¹⁸ this led to an improved 47% isolated yield and 2.7:1 C6:C5 regioselectivity (entry 6). Once it was judged that dtbpy (L1) had reached its limit in terms of regioselectivity, the ligand was changed. An immediate effect was observed when using tetramethylphenanthroline (Me₄phen; L2),¹⁹ whereby 64%isolated yield of a 3.8:1 mixture of 6 and 5 was obtained (entry 7). Eliminating HBPin led to lower yield (entry 8), whereas removing B_2Pin_2 was detrimental to the reaction (entry 9). Although a longer reaction time did not improve the reaction (entry 10), slightly higher temperatures led to 100% conversion (74% isolated yield) for the first time (entry 11). Using these optimized conditions of 5 mol% [Ir(cod)OMe]₂, 10 mol% ligand, 0.25 equiv of HBPin, and 4.0 equiv of B₂Pin₂ in hexanes at 80 °C in a sealed tube for 24 h, a further ligand screen was conducted (entries 12-14). Out of these, phenanthroline (phen; L3) was the most effective, resulting in 77% isolated yield and 8.0:1 C6:C5 regioselectivity (entry 12). Since reduction of catalyst loading to 2 mol% slightly decreased the yield (entry 15), the conditions from entry 12 were retained.

Although this optimization was sufficient for the initial goal of generating 6-substituted tryptophans, a substrate screen was performed to examine the scope of N,C3-disubstituted indole substrates that can undergo C6–H borylation (Figure 2B). Rescreening ligands L1–L3 for every substrate revealed that phen (L3) was indeed the best ligand for this transformation. Both C3-substituted and C2,C3-disubstituted indoles were viable substrates (7–11); however, sensitive functional groups such as a primary bromide (as in 12) were not tolerated. Of

note, reaction of desilylated 4 (i.e., Boc-L-Trp-OMe) led to only small amounts of C2- and C7-borylated products (not shown), indicating the importance of the TIPS group.^{15c,d} As an extension of this indole reaction methodology, *N*-TIPS-carbazole (13) was reacted to give C2,C7-bis-borylated product, and C3-ethylated carbazole 14 provided C7-monoborylated product.

With this enabling method in hand, the total synthesis of fumitremorgin A (1) and vertuculogen (2) from inexpensive Ltryptophan became a more realistic endeavor since decagram quantities of 3 were now accessible. This undertaking, however, was not without further challenges (Figure 3A).²⁰ Thus, a variety of dihydro- and tetrahydro- β -carbolines, represented by the general structure 15, were subjected to oxidants in an attempt to realize a late-stage oxidation that mimics the biosynthetic pathway to 1 and 2.²¹ In spite of these efforts, none of these reactions resulted in the endoperoxide 16. Some of these reactions, however, encouragingly led to intermediates such as 17 and 18 that contain the required peroxide and cyclic motifs (Figure 3B). Studying these systems further, we eventually arrived at hydroperoxides 19a and 19b, whose NH and OOH groups can be cyclized onto 3-methyl-2-butenal (20) using $BF_3 \cdot OEt_2$ (Figure 3C). The presence of the eightmembered endoperoxide (as opposed to a seven-membered ether like 18) and the diastereoselectivity of the hemiaminal stereocenter were confirmed by X-ray analysis of the starting material 19a and the product 21a. Notably, 21a and 21b contain the requisite motifs of the oxidizing peroxide bond and oxidizable π bonds present in alkaloids 1 and 2.

This groundwork ultimately paved the way to the first total synthesis of fumitremorgin A (1) and verruculogen (2; Figure 3D). Commercially available Boc-L-Trp-OMe (22) was protected with TIPS-Cl to give the precursor for the Ir-catalyzed borylation, Boc-L-Trp(TIPS)-OMe (4). As described in Figure 2, 4 was borylated at the C6 position and then immediately subjected to Chan–Lam coupling²² with methanol



^aReagents and conditions: (a) LiHMDS, TIPS-Cl, THF, -78 °C (89%); (b) Ir[(cod)OMe]₂ (5 mol %), phenanthroline (10 mol%), HBPin (0.25 equiv), B₂Pin₂ (4.0 equiv), hexanes, 80 °C, 24 h; then Cu(OAc)₂, Et₃N, MeOH, O₂, 23 °C, 24 h (65%, 8:1 C6:C5); (c) 3 N HCl, MeOH, 60 °C; (d) 1 M TBAF, THF (91% over 2 steps); (e) **24**, 4Å MS, CHCl₃, 0 °C, 1 h; then TFA (10 mol%), 0 °C, 24 h (49%, 2:1 dr desired:undesired); (f) **26**, CH₂Cl₂, sat. aq. NaHCO₃ (87%); (g) PhNO, ZrCl₄, CH₂Cl₂, 0 °C (76%); (h) Et₂NH, THF, 40 °C (91%); (i) OsO₄ (2.5% in 'BuOH), *N*-methylmorpholine *N*-oxide (50% in H₂O), MeCN:acetone:H₂O (2:2:1) (79%); (j) 1 M TBAF, THF, ACOH, DMF, 0 °C; (k) **20**, 4Å MS, BF₃•OEt₂, DCE, -20 °C (30% over 2 steps); (l) prenyl bromide, Bu₂SnO, Bu₄NI, DCE, 23 °C, 17 h (95%).

Figure 3. Total synthesis of vertuculogen (2) and fumitremorgin A (1). (A) Failed attempts at directly oxidizing the indole core 15. (B) Intermediates isolated while attempting to forge the eight-membered endoperoxide. (C) Successful eight-membered endoperoxide formation. (D) First total synthesis of vertuculogen (2) and fumitremorgin A (1) enabled by indole C6-borylation/oxidation and eight-membered endoperoxide closure.

in one pot to give 23 in 65% yield and with 8:1 C6:C5 selectivity. Deprotections with HCl and TBAF led to 6methoxytryptophan methyl ester (3; not shown) in 91% yield from 23. Pictet-Spengler reaction with TBDPS-protected peroxy-aldehyde 24 (see the Supporting Information for its preparation from 20) led to tricyclic peroxide 25 in 49% yield (2:1 dr). Coupling with N-Fmoc-L-prolyl chloride (26) then gave 27 in 87% yield. Late-stage dehydrogenation with $ZrCl_4$ PhNO^{9d} proved essential, as other oxidants such as DDQ failed in this transformation. Fmoc removal gave pentacycle 28 (69% yield over two steps), followed by a chemoselective dihydroxylation with OsO₄ to give **29** (79% yield).⁶ Treatment with TBAF allowed the removal of the TBDPS group, followed by the crucial endoperoxide-forming cyclization with aldehyde 20 to give vertuculogen (2) in 30% yield over two steps. It is of note that this final annulation reaction was achieved with complete diastereoselectivity. Finally, prenylation of 2 using prenvl bromide, much like the biosynthetic pathway with prenyl pyrophosphate,²¹ led to fumitremorgin \hat{A} (1) in 95% vield.

The longstanding synthetic challenge posed by the peroxidecontaining alkaloids,²³ vertuculogen (2) and fumitremorgin A (1), has thus been put to rest in 11 and 12 steps, respectively, starting from a commercially available tryptophan derivative. The enabling regioselective C–H borylation of the remote C6 position of tryptophan is a forceful reminder of the simplifying power of such disconnections.²⁴ One potential application of this work is the synthesis of valuable unnatural amino acids through C–H borylation and subsequent derivatization.^{15d} Lastly, this method was conducted on a variety of related indole systems and, not surprisingly, has already been field-tested on decagram scale at Novartis for incorporation into a medicinal chemistry program.²⁵

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b07154.

¹H and ¹³C NMR spectra for all new compounds (PDF)

Experimental procedures and characterization data (PDF)

X-ray crystallographic data for 19 (CIF)

X-ray crystallographic data for 21 (CIF)

AUTHOR INFORMATION

Corresponding Author

*pbaran@scripps.edu

Author Contributions

[‡]Y.F. and D.H. contributed equally to this paper.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

Financial support for this work was provided by NIH/NIGMS (GM-073949), NSF (predoctoral fellowship to D.H.), DAAD (predoctoral fellowship to J.Z.), and JSPS (postdoctoral fellowship to S.U.). The authors thank Prof. A. L. Rheingold and Dr. C. E. Moore for X-ray crystallographic analysis, Dr. W. Gutekunst for assistance in the early phases of this project, and Dr. Y. Ishihara for assistance in the preparation of this manuscript.

REFERENCES

(1) Isolation and/or structural determination of fumitremorgin A (1): (a) Yamazaki, M.; Suzuki, S.; Miyaki, K. *Chem. Pharm. Bull.* **1971**, *19*, 1739. (b) Yamazaki, M.; Fujimoto, H.; Kawasaki, T. *Chem. Pharm. Bull.* **1980**, *28*, 245.

(2) Isolation and/or structural determination of verruculogen (2):
(a) Cole, R. J.; Kirksey, J. W.; Moore, J. H.; Blankenship, B. R.; Diener, U. L.; Davis, N. D. Appl. Microbiol. 1972, 24, 248. (b) Cole, R. J.; Kirksey, J. W. J. Agric. Food Chem. 1973, 21, 927. (c) Fayos, J.; Lokensgard, D.; Clardy, J.; Cole, R. J.; Kirksey, J. W. J. Am. Chem. Soc. 1974, 96, 6785. (d) Cole, R. J.; Kirksey, J. W.; Cox, R. H.; Clardy, J. J. Agric. Food Chem. 1975, 23, 1015. (e) Yoshizawa, T.; Morooka, N.; Sawada, Y.; Udagawa, S. Appl. Environ. Microbiol. 1976, 32, 441.

(3) Liu, D.-Z.; Liu, J.-K. Nat. Prod. Bioprospect. 2013, 3, 161.

(4) (a) Rabindran, S. K.; Ross, D. D.; Doyle, L. A.; Yang, W.; Greenberger, L. M. *Cancer Res.* **2000**, *60*, 47. (b) Allen, J. D.; van Loevezijn, A.; Lakhai, J. M.; van der Valk, M.; van Tellingen, O.; Reid, G.; Schellens, J. H. M.; Koomen, G.-J.; Schinkel, A. H. *Mol. Cancer Ther.* **2002**, *1*, 417.

(5) Wang, X.; Furukawa, T.; Nitanda, T.; Okamoto, M.; Sugimoto, Y.; Akiyama, S.-I.; Baba, M. *Mol. Pharmacol.* **2003**, *63*, 65.

(6) For a review of the synthesis of fumitremorgins and verruculogen-TR2, see: Hino, T.; Nakagawa, M. *Heterocycles* **1997**, 46, 673.

(7) (a) Williams, R. M.; Stocking, E. M.; Sanz-Cervera, J. F. Top. Curr. Chem. 2000, 209, 97. Also see the following reviews on indole alkaloids, as well as their preceding compilations: (b) Saxton, J. E. Nat. Prod. Rep. 1997, 14, 559. (c) Toyota, M.; Ihara, M. Nat. Prod. Rep. 1998, 15, 327. (d) Ishikura, M.; Abe, T.; Choshi, T.; Hibino, S. Nat. Prod. Rep. 2013, 30, 694.

(8) Taniguchi, M.; Anjiki, T.; Nakagawa, M.; Hino, T. Chem. Pharm. Bull. 1984, 32, 2544.

(9) (a) Gan, T.; Liu, R.; Yu, P.; Zhao, S.; Cook, J. M. J. Org. Chem. 1997, 62, 9298. (b) Ma, C.; Liu, X.; Li, X.; Flippen-Anderson, J.; Yu, S.; Cook, J. M. J. Org. Chem. 2001, 66, 4525. (c) Baran, P. S.; Guerrero, C. A.; Ambhaikar, N. B.; Hafensteiner, B. D. Angew. Chem., Int. Ed. 2005, 44, 606. (d) Baran, P. S.; Hafensteiner, B. D.; Ambhaikar, N. B.; Guerrero, C. A.; Gallagher, J. D. J. Am. Chem. Soc. 2006, 128, 8678. (e) Jia, Y.; Zhu, J. Synlett 2005, 2469.

(10) Although 6-methoxyindole is commercially available, it is costly (\$306/gram) and is significantly more expensive than L-tryptophan methyl ester (\$5.26/gram); 6-hydroxyindole is not much cheaper (\$240/gram) [Sigma-Aldrich 2015 prices].

(11) (a) Harvey, D. G.; Robson, W. J. Chem. Soc. 1938, 97.
(b) Bergmann, E. D.; Hoffmann, E. J. Chem. Soc. 1962, 2827.
(c) Allen, M. S.; Hamaker, L. K.; La Loggia, A. J.; Cook, J. M. Synth. Commun. 1992, 22, 2077. (d) Blaser, G.; Sanderson, J. M.; Batsanov, A. S.; Howard, J. A. K. Tetrahedron Lett. 2008, 49, 2795.

(12) (a) Kermack, W. O.; Perkin, W. H.; Robinson, R. J. Chem. Soc., Trans. **1921**, 119, 1602. (b) Feldman, P. L.; Rapoport, H. Synthesis **1986**, 1986, 735.

(13) Gutekunst, W. R.; Baran, P. S. Chem. Soc. Rev. 2011, 40, 1976.
(14) (a) Takagi, J.; Sato, K.; Hartwig, J. F.; Ishiyama, T.; Miyaura, N. Tetrahedron Lett. 2002, 43, 5649. (b) Kallepalli, V. A.; Shi, F.; Paul, S.; Onyeozili, E. N.; Maleczka, R. E.; Smith, M. R. J. Org. Chem. 2009, 74, 9199. (c) Kawamorita, S.; Ohmiya, H.; Sawamura, M. J. Org. Chem. 2010, 75, 3855.

(15) (a) Paul, S.; Chotana, G. A.; Holmes, D.; Reichle, R. C.; Maleczka, R. E.; Smith, M. R. J. Am. Chem. Soc. 2006, 128, 15552.
(b) Robbins, D. W.; Boebel, T. A.; Hartwig, J. F. J. Am. Chem. Soc. 2010, 132, 4068. (c) Homer, J. A.; Sperry, J. Tetrahedron Lett. 2014, 55, 5798. (d) Loach, R. P.; Fenton, O. S.; Amaike, K.; Siegel, D. S.; Ozkal, E.; Movassaghi, M. J. Org. Chem. 2014, 79, 11254.

(16) Ishihara, Y.; Montero, A.; Baran, P. S. The Portable Chemist's Consultant: A Survival Guide for Discovery, Process, and Radiolabeling; Apple Publishing Group: Cupertino, CA, 2015.

(17) Saito, Y.; Segawa, Y.; Itami, K. J. Am. Chem. Soc. 2015, 137, 5193.

(18) (a) Boller, T. M.; Murphy, J. M.; Hapke, M.; Ishiyama, T.; Miyaura, N.; Hartwig, J. F. J. Am. Chem. Soc. 2005, 127, 14263.
(b) Preshlock, S. M.; Ghaffari, B.; Maligres, P. E.; Krska, S. W.; Maleczka, R. E.; Smith, M. R. J. Am. Chem. Soc. 2013, 135, 7572.

(19) (a) Liskey, C. W.; Hartwig, J. F. J. Am. Chem. Soc. 2012, 134, 12422. (b) Larsen, M. A.; Hartwig, J. F. J. Am. Chem. Soc. 2014, 136, 4287.

(20) Holte, D. Part I. Synthetic studies in sesquiterpenes; Part II. Total synthesis of verruculogen. Ph. D. thesis, The Scripps Research Institute, La Jolla, CA, 2014.

(21) Mundt, K.; Wollinsky, B.; Ruan, H.-L.; Zhu, T.; Li, S.-M. ChemBioChem 2012, 13, 2583.

(22) Shade, R. E.; Hyde, A. M.; Olsen, J.-C.; Merlic, C. A. J. Am. Chem. Soc. 2010, 132, 1202.

(23) For recent syntheses of peroxide-containing terpenes, see: (a) Cook, S. P. Synlett **2014**, 25, 751. (b) Hu, X.; Maimone, T. J. J. Am. Chem. Soc. **2014**, 136, 5287.

(24) Fischer, D. F.; Sarpong, R. J. Am. Chem. Soc. 2010, 132, 5926.
(25) Personal communication with Dr. Tetsuo Uno at the Genomics Institute of the Novartis Research Foundation.